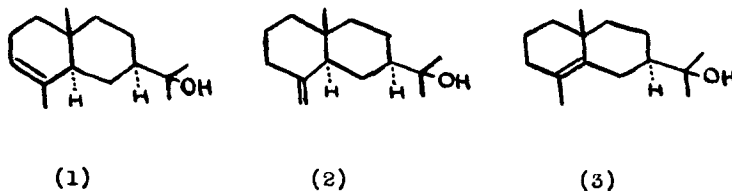


SYNTHESIS OF (+)- β -EUDESMOL

R. P. Houghton, D. C. Humber and *A. R. Pinder
Department of Chemistry, University College, Cardiff
(University of Wales)

(Received 29 November 1965)

Eudesmol has long been recognised as being a mixture of two isomers, α - and β -, formulated on degradative evidence as (1) and (2) respectively.¹ A third isomer, γ -eudesmol (3), is also present in some samples of commercial eudesmol.²

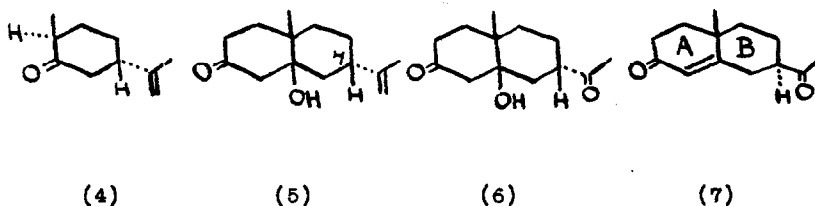


The constitution assigned to γ -eudesmol has been confirmed by total synthesis.³ The relative stereochemistry (2) for β -eudesmol has been deduced from chemical studies,⁴ and this has been confirmed recently by an elegant stereoselective synthesis of racemic β -eudesmol.⁵ It has also been proved, by correlation with intermediates in steroid

* Present address: Department of Chemistry, Clemson University, Clemson, S. Carolina, 29631, U.S.A.

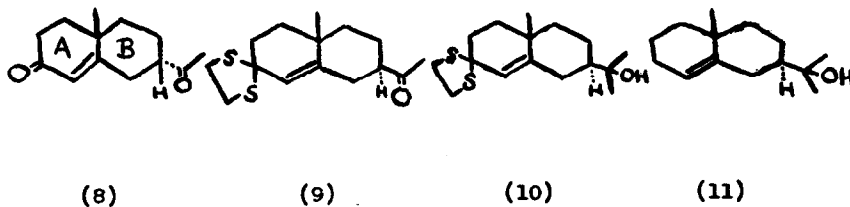
synthesis of known absolute configuration, that (2) represents (+)- β -eudesmol.⁶

We describe here a total synthesis of (+)- β -eudesmol by a stereoselective route. A Robinson-Mannich condensation between (-)-dihydrocarvone, of established stereochemistry (4),⁷ and nascent methyl vinyl ketone (from 4-diethylaminobutan-2-one methiodide and methylsulphuryl carbanion in dimethyl sulphoxide)⁸ afforded the cis-2-decalone (5), the stereochemistry at the ring junction being assigned from a knowledge of the steric course of the reaction.⁹ ORD measurements

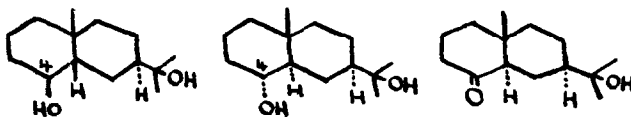


on (5) indicated the rings were fused in the non-steroidal conformation,¹⁰ further evidence for this being the chemical shift ($\delta = 1.18$ ppm, relative to TMS = 0) of the angular methyl protons in the N.M.R. spectrum of (5).¹¹ Ozonolysis of (5) in chloroform led to the diketo-alcohol (6), which on acid dehydration afforded the octalone (7). During the dehydration step an inversion occurred at C₍₇₎,¹² for the ORD curve of the product was very similar to that of cholest-4-en-3-one¹³ and quite different from those of epi- α -cyperone¹³ and 8-isotestosterone,¹³ suggesting that ring B in (7) is a chair, with the acetyl substituent equatorial and β . If ring B of

the uninverted structure (8) were a chair the acetyl group would be axial, but it is almost certain that this ring is a boat with the C₍₇₎-substituent boat-equatorial (cf. epi- α -cyperone¹⁴). The correctness



of this configurational assignment was corroborated by the subsequent steps in the synthesis. Reaction of (7) with ethanedithiol resulted in a preferential attack¹⁵ of the conjugated keto-group with the formation of the monodithioketal (9), the ORD curve of which was closely similar to that of cholest-4-en-3-one dithioketal.¹⁶ Treatment of (9) with magnesium methyl iodide gave the tertiary alcohol (10), which on desulphurisation with sodium and liquid ammonia¹⁷ afforded nor- γ -eudesmol (11). The ORD curve of this product was closely similar to that of γ -eudesmol (3).³ Hydroboration of (11) under carefully defined conditions¹⁸ yielded a mixture of the diols (12) and (13), which was oxidised directly with Jones reagent¹⁹ to a mixture of the corresponding 4-ketones, homogenised²⁰ on basic alumina to the crystalline trans ketol (14). This compound proved to be identical in every respect with the ketol obtained by ozonolysis of



(12)

(13)

(14)

commercial eudesmol.^{4,21} Finally, a Wittig reaction between (14) and methylenetriphenylphosphorane in dimethyl sulphoxide solution²² afforded (+)- β -eudesmol, purified by chromatography and sublimation, and identical in all respects with natural (+)- β -eudesmol (2).

This synthesis is also a total synthesis of cryptomeridiol²³ and neointermedeol,²⁴ which have been synthesised from (+)- β -eudesmol.

All the new compounds described have satisfactory spectral properties and analyses.

Acknowledgements:

We acknowledge with thanks the award of a British Nylon Spinners, Ltd. research studentship (to D.C.H.), and the Government Grants Committee of the Royal Society for a research grant. We thank Messrs. Plaimar Ltd., Perth, Western Australia, for a gift of commercial eudesmol, and Professor S. C. Bhattacharyya, Poona, India, for a sample of pure (+)- β -eudesmol.

REFERENCES

1. For a review of the chemistry of eudesmol see D. H. R. Barton in "Chemistry of Carbon Compounds", ed. E. H. Rodd, Elsevier, Amsterdam, 1953, Vol. IIB, p. 664.
2. R. B. Bates and E. K. Hendrickson, Chem. and Ind., 1759 (1962);
M. L. Seikel and J. W. Rowe, Phytochemistry, 3, 27 (1964);
E. von Rudloff, Chem. and Ind., 2126 (1964).
3. A. R. Pinder and R. A. Williams, J. Chem. Soc., 2773 (1963).
4. A. J. Birch and K. M. C. Mostyn, Austral. J. Chem., 7, 301 (1954).
5. J. A. Marshall and M. T. Pike, Tetrahedron Letters, 3107 (1965).
6. B. Riniker, J. Kalvoda, D. Arigoni, A. Fürst, O. Jeger, A. M. Gold and R. B. Woodward, J. Am. Chem. Soc., 76, 314 (1954).
7. A. J. Birch, Ann. Reports Chem. Soc., 47, 191 (1950), and references there cited.
8. E. J. Corey and M. Chaykovsky, J. Am. Chem. Soc., 84, 866 (1962);
G. G. Price and M. C. Whiting, Chem. and Ind., 775 (1963).
9. F. J. McQuillin, J. Chem. Soc., 528 (1955); T. A. Spencer,
K. K. Schmiegel and K. L. Williamson, J. Am. Chem. Soc., 85, 3785
(1963); J. A. Marshall and W. I. Fanta, J. Org. Chem., 29, 2501 (1964).
10. We are greatly indebted to Professor W. Klyne, Dr. P. M. Scopes and Dr. S. Wallis (Westfield College, University of London) for ORD measurements and discussions.
11. W. G. Dauben, R. M. Coates, N. D. Vietmeyer, L. J. Durham and C. Djerassi, Experientia, 21, 565 (1965).

12. Compare T. G. Halsall and D. W. Theobald, J. Chem. Soc., 1029 (1964).
13. C. Djerassi, R. Riniker and B. Riniker, J. Am. Chem. Soc., 78, 6377 (1956).
14. F. J. McQuillin, J. Chem. Soc., 528 (1955); C. Djerassi, R. Riniker and B. Riniker, J. Am. Chem. Soc., 78, 6362 (1956).
15. J. W. Ralls and B. Riegel, J. Am. Chem. Soc., 76, 4479 (1954).
16. D. A. Lightner, C. Djerassi, K. Takeda, K. Kuriyama and T. Komeno, Tetrahedron, 21, 1581 (1965).
17. R. E. Ireland, T. I. Wrigley and W. G. Young, J. Am. Chem. Soc., 80, 4604 (1958).
18. H. C. Brown and G. Zweifel, J. Am. Chem. Soc., 83, 2544 (1961).
19. K. Bowden, I. M. Heilbron, E. R. H. Jones and B. C. L. Weedon, J. Chem. Soc., 39, (1946).
20. M. Nussim, Y. Mazur and F. Sondheimer, J. Org. Chem., 29, 1120 (1964).
21. I. Ruzicka, A. H. Wind and D. R. Koolhaas, Helv. Chim. Acta, 14, 1132 (1931).
22. R. Greenwald, M. Chaykovsky and E. J. Corey, J. Org. Chem., 28, 1128 (1963).
23. M. Sumimoto, H. Ito, H. Hirai and K. Wada, Chem. and Ind., 780 (1963).
24. V. B. Zalkow, A. M. Shaligram and L. H. Zalkow, Chem. and Ind., 194. (1964).